

Is sodium sulfate decahydrate a phase change energy storage material?

In this paper, sodium sulfate decahydrate (SSD) with a phase transition temperature of 32 °C was selected as the phase change energy storage material. However, SSD has the problems of large degree of supercooling, obvious phase stratification, and low thermal conductivity.

Is sodium sulfate a good salt hydrate?

For example, sodium sulfate decahydrate, $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ (SSD), has been identified as one of the most promising salt hydrates for building applications due to its low cost (1.60 \$/kWh), high energy storage capacity (254 J/g), and moderate melting temperature (32.4 °C) [20,21].

Can salt hydrates be used for thermal storage in buildings?

Limited by temperature range for most salt hydrates, using salt hydrates for thermal storage in buildings have always attracted interests of researchers due to their abilities for changing the increasingly indifferent energy generation. It can be deduced the trends in decreasing the energy demand of buildings is necessary.

What is the melting temperature of sodium sulfate?

The melting temperature of sodium sulfate in the composite materials is around 880 °C and no confinement effect is observed due to the nanoscale diatomite pore size. This indicates that the material should be used at temperatures over 890 °C to maximize the energy density.

Are salt hydrates suitable for latent heat storage?

It is generally agreed that salt hydrates have significant merits among phase change materials for latent heat storage. In spite of some disadvantages may limit their application in certain aspects, corresponding solutions are developed continuously.

Are salt hydrates a good choice for building applications?

However, with a typical material cost of 20-40 \$/kWh, they are too expensive for most building applications [16]. On the contrary, salt hydrates are promising candidates because of their low cost and high thermal energy storage density [.,].

Sodium sulfate decahydrate ($\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$) is known to decompose peritectically upon heating to 32.4 °C to yield anhydrous sodium sulfate and a saturated solution of Na_2SO_4 in water. ...

The calcium hexaaluminate was used as a novel skeleton material to develop shape-stabilized sodium sulfate. The composites have an energy density of 1861 J/g and a ...

A selection and optimization experimental study of additives to thermal energy storage material sodium acetate trihydrate. In Proceedings of the International Conference on Energy and Environment Technology,

Guilin, China, 16-18 ...

?: (sodium sulfate decahydrate,SSD) (2.4 ?) (200 J/g) ...

A way to overcome issues related to the exploitation of solar energy is to refer to concentrated solar power technology coupled with systems for thermochemical energy storage (TCES) as a ...

In this study, two compositions in the system Li_2SO_4 - Na_2SO_4 are considered as candidate materials for thermal energy storage at high temperatures (450-550 °C): 79/21 and 50/50 ...

Study on performance optimization of sodium sulfate decahydrate phase change energy storage materials Xian Dong¹ · Jinfeng Mao¹ · Shibin Geng¹ · Yong Li¹ · Pumin Hou¹ · Huiliang ...

Journal Article: Thermal energy storage using sodium sulfate decahydrate and water ... It appears that the major problem preventing use of sodium sulfate decahydrate for thermal energy ...

Sodium sulfate decahydrate ($\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$), also known as mirabilite or Glauber's salt, has been the most investigated salt hydrate for use in latent thermal energy ...

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