

When X-H are near another Y atom (proton acceptor) with high electronegativity (O, N, F), the molecular structure of X-H...Y will spontaneously form, resulting in hydrogen ...

This performance stems from abundant quinone/imine functionalities that decorate an extended aromatic backbone, act as redox-active sites, engage in hydrogen bonding, and enable a delocalized high-rate energy ...

The oxidative addition of the H-H bond to a single main-group centre in a metallomimetic fashion and its further use in catalysis is challenging. Now the oxidative addition of the H-H bond to ...

Then it provided the external electric field about ~ 2 V/° as a solution to improve the desorption temperature to 668 K by enhancing the average adsorption energy. And the ...

High reliability and proven ultra-long life make aqueous batteries ideal for grid energy storage. However, the narrow electrochemical stability window (ESW) caused by the high activity of H₂O severely hampers their ...

Both non-renewable energy sources like coal, natural gas, and nuclear power as well as renewable energy sources like hydro, wind, wave, solar, biomass, and geothermal energy can be used to produce hydrogen. The ...

To break the strong hydrogen-bond network, an additional energy of 25.44 kJ mol⁻¹ is needed, which makes it more difficult for water molecules to adsorb and deprotonate on ...

Aqueous batteries are emerging as highly promising contenders for large-scale grid energy storage because of uncomplicated assembly, exceptional safety, and cost-effectiveness. The unique aqueous electrolyte ...

The fixed orientation of hydrogen bonds is significantly weaker than that of covalent and coordination bonds. The bond angles of stronger hydrogen bonds can also vary ...

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